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AT BE CH DE ES FR GB GR IT LI LU NL SE(71) Applicant: X-Flow B.V.  
Hengelsestraat 705  
NL-7521 PA Enschede(NL)Applicant: PRIMECARE B.V.  
Torenlaan 55  
7559 PK Hengelo(NL)(72) Inventor: Koenhen, Dirk Marinus  
Hengelsestraat 705  
NL-7521 PA Enschede(NL)  
Inventor: Scharstuhl, Johan Jan  
Torenlaan 55  
NL-7559 PK Hengelo(NL)(74) Representative: Kupecz, Arpad et al  
Octrooibureau Los en Stijger B.V. Postbox  
20052  
NL-1000 HB Amsterdam(NL)

(54) A process and device for the separation of a body fluid from particulate materials in said fluid, in particular plasma from whole blood, to be used for a chemical analysis and testing kit for said separation and analysis.

(57) A process and device for the separation of a body fluid from particulate materials in said fluid to be used for a chemical analysis of the components present therein. Said process and device are in particular suitable for the separation of plasma from whole blood.

Furthermore the present invention relates to a testing kit for the above mentioned separation and analysis purposes.

Because of the simplicity of the present process and device as well as the testing kit a routine examination may be carried out not only in clinical laboratories but also by general practitioners.

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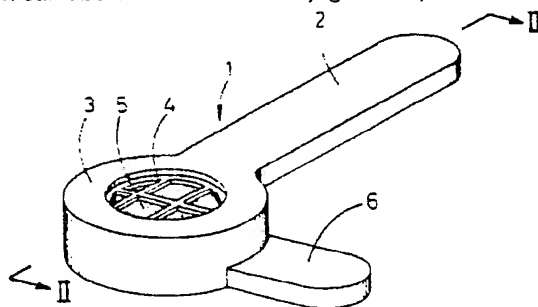


fig.1

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**A process and device for the separation of a body fluid from particulate materials in said fluid, in particular plasma from whole blood, to be used for a chemical analysis and testing kit for said separation and analysis.**

The present invention relates to a process for the separation of a body fluid from the particulate materials present in said fluid, in particular plasma from whole blood, to be used for a chemical analysis wherein a small amount of body fluid is contacted with a membrane or composite of membranes, said body fluid migrates into said membrane or composite whereas said particulate materials and non-migrated body fluid are retained and said migrated fluid is used for analysis purposes. Furthermore the present invention relates to a device for the separation of a body fluid from the particulate materials present in said fluid, in particular plasma from whole blood, comprising an inert solid support to which a membrane or composite of membranes is attached. Finally the invention relates to a testing kit for separation and chemical analysis of the components of a body fluid.

Although the present invention relates to the separation of a body fluid from the particulate materials present therein and to a chemical analysis of said body fluid in general, such as lymph, urine and other body fluids the present invention will be further disclosed and illustrated by means of whole blood as a body fluid.

In view of diagnostic and therapy and for controlling of certain body functions there is since many years a need for analysis of body fluids in order to analyse the presence and/or concentration of a certain substance in said fluid. In particular blood is a frequently examined medium, which may lead to indications for several body functions and body processes. Since blood is a troubled medium because of the several particulate materials present therein, such as thrombocytes (blood platelets), erythrocytes (red blood cells) and leucocytes (white blood cells) whole blood is a troublesome medium for chemical analysis, since such particulate materials do interfere with the usual analytical chemical techniques used for the determination of the presence and concentration of certain components. Accordingly several methods have been developed for the separation on the one hand of particulate materials and on the other hand of the plasma or serum from whole blood. For the somewhat larger amounts of blood (in the order of millimeters and more) many methods have been developed, which are mostly based on the centrifugation principle, wherein the blood cells are precipitated from the plasma or serum by means of the centrifugal force. However such methods usually do require complicated adjustments in order to recover the plasma or serum without the particulate materials. Another disadvantage of such techniques is that they require greater amounts of blood, which should be taken off by means of a venous puncture. Because of the above mentioned facts such methods are only used in the clinical chemical laboratory.

By the increasing demand for analysis data of body fluids, in particular blood, and by the increasing need for quick and simple analysis methods means are developed which only require a slight amount of blood, f.i. a drop of blood, obtained via a finger puncture. Such means exist usually in the form of a strip of an inert material, to which several layers are applied, which are able to achieve a separation between the particulate materials of the blood on the one hand and plasma or serum on the other hand, so that the particulate materials are retained in or on one or several of the upper layers, whereas the clear plasma is collected in the layers thereunder. In one or several of these layers reagents may be introduced. These reagents do simultaneously or after each other get in contact with the plasma and therefore they may react with the component to be determined. Because of this reaction a change in a physical property does occur, which change is a measure for the concentration of the substance to be determined, which change is observed visually or by means of a suitable device, which leads to a semi-quantitative and quantitative result respectively.

Accordingly the U.S. patent specification 3,092,465 discloses the coating of a strongly sucking test paper, which is provided with a semi-permeable membrane for the determination of the blood sugar content. Herein blood is applied to the semi-permeable membrane, through which membrane only water and glucose pass but other bigger molecules such as haemoglobine or proteins are retained. Said glucose and water are taken up in the paper layer, which is provided with suitable reagents. The changing colour such obtained is then measured visually or remission- or reflection photometrically. This change in colour is observed through the semi-permeable membrane, which has therefore to be transparent and from which the disturbing particulate materials should be removed by means of washing or wiping. Such a method is only suitable for the determination of substances having small molecules.

Further it is known from German "Offenlegungsschrift" 1,598,153 to use a film which may swell up in water, which contains reagents. The components which are dissolved in the blood penetrate into the semi-permeable film, whereas the particulate materials are retained. This method has however similar disadvantages.

tages as the method which is disclosed in U.S. patent specification 3,092,465.

In order to improve such systems having semi-permeable membranes or films, several systems have been developed in order to separate particulate materials from blood to recover plasma or serum. Accordingly German "Offenlegungsschrift" 2,222,951 discloses a testing device for the determination of the enzyme activity in blood, wherein several different layers are arranged after each other, whereas one or several of these layers do function as a filter. Herein the upper two layers are porous glass fibre discs, which act as prefilter in order to retain particulate materials such as white blood cells, in order to prevent clogging of the underlying membrane filter disc. The membrane filter disc acts as a filter for removing the red blood cells and other particles. Finally under said membrane filter disc there is provided a disc of cellulose acetate. Each of the separate discs provide distinctive test zones. The colour change which is obtained by such analysis method is observed visually. The disadvantage of such a method is that the plasma only very slowly and in slight amounts may penetrate into the membrane filter, whereas this membrane filter may be clogged easily, which means that said plasma slowly and unequally penetrates into the reaction layer. Furthermore German "Offenlegungsschrift" 2,922,958 discloses a multi layer element, having at least four layers consisting of an upper filter layer, a water impermeable layer thereunder having one or more orifices, then an expanding layer and finally a reagent layer. As filter layer use is made of one or several membrane filters, whereas the porous expanding layer may also be a membrane filter. It is stated that the expanding layer is weak and fragile (see page 15, sub 5) so that this layer should be sandwiched between other layers in order to prevent problems during handling. Such a multi layer element has similar disadvantages as mentioned in the German "Offenlegungsschrift" 2,222,951.

The above discussed methods wherein on the one hand use is made of semi-permeable membranes and on the other hand use is made of membrane filters (porous membranes) did not lead to fully satisfactory results. A further development of such methods is disclosed in the European patent application 45476, wherein it is suggested to apply a layer of glass fibres in the form of a disc onto the known testing devices as cover. The glass fibre disc retains the particulate materials from the blood and enables to pass serum or plasma into the layers thereunder. As disadvantages of such a system are mentioned (see European patent application 133895, page 3) the relatively high dead volume of this glass fibre disc as well as the fact that such a disc is not able to retain the erythrocytes completely, so that blood colour penetrates into the reaction zone which disturbs the reaction. Therefore in the last mentioned European patent application a substrate is suggested which should be contacted with blood and which substrate together with blood should be passed through a specific filter disc, so that the separated plasma may be recovered in the layers thereunder.

Finally according to the methods disclosed in the European patent applications Nrs. 0154839 and 0175990 the blood sample is applied to that side of the membrane where the pore size is the smallest (the so-called smooth surface of the membrane). This has as serious disadvantage that the small pores are clogged by the blood particles such as thrombocytes, erythrocytes, leucocytes and coagulates thereof, resulting in the fact that only very slight amounts of plasma penetrate into the membrane. This means that the reaction time is too long with the risk that the determination is inaccurate. (In connection with the above mentioned disadvantages reference is made to European patent application No. 0045476 in the name of Boehringer Mannheim GmbH, page 2, lines 14 through 20).

Another disadvantage is that the blood has to be wiped off from the smooth surface of the membrane, whereby damaging of the erythrocytes is practically unavoidable, which means that the desired colour reaction is disturbed.

In conclusion there are several methods and testing devices available, which are also known as quick diagnostics, which may be used for the analysis of a slight amount, f.i. a drop, of whole blood. However in case of the use of semi-permeable membranes there is not obtained the total plasma or serum, but only the components to be determined with a small molecule size, whereas in using membrane filters usually complicated provisions should be taken in order to safeguard a proper process. Furthermore it should be noted that frequently cloggings occur and that the plasma or serum only slowly passes through such membranes. The recent developments in this art make use of other filter discs than membrane filters. Except the above mentioned disadvantages and shortcomings each of the above mentioned methods relate to "dry" analysis methods. This means that the result, usually a colour change, should be evaluated visually or by means of reflection photometric. A visual method is however not quantitative, whereas a reflection photometric method requires a relatively expensive measuring device and makes high demands of the test systems. Further the known test systems are usually provided with swellable reagent layers, so that it is often difficult to obtain a reproduceable result, since a drop of blood, in particular when said drop of blood should be applied on such a test system, mostly does not lead to a reproduceable result. Furthermore the long time period which is required for the plasma to penetrate in such systems may result in the fact that

the blood begins to coagulate, resulting in clogging of the entire system.

Therefore there is a need for a simple and cheap method and device for the chemical analysis of components of body fluids, wherein use is made of a drop or a small amount of such a liquid, wherein the above mentioned disadvantages are eliminated. Furthermore there is also need for a quick, simple and accurate "wet" analysis of such fluids, wherein relatively cheap analysis apparatus may be used, such as f.i. colorimeters or photometers, which means that such a method is accessible for routine examinations, not only in clinical laboratories, but also by general practitioners and so on.

Surprisingly it has been found that by making use of special membranes said goals may be achieved. Therefore the process for the separation of a body fluid from the particulate materials in said fluid, in particular plasma from blood, by contacting a slight amount of body fluid with a membrane or composite of membranes, wherein the body fluid penetrates into the membrane or composite under retaining of any particulate materials and not-penetrated body fluid, wherein the penetrated fluid is used for analysis purposes, according to the invention is characterized in that one or more hydrophilic microporous membranes, at least one of which (the collector membrane) has a defined pore volume, is or are used, wherein the absorbed, defined amount of body fluid which is free of the particulate material, is used for a semi-quantitative or quantitative, dry or wet chemical analysis in order to determine one or more of its components.

By treating in this way a slight amount of body fluid, e.g. a drop, one achieves quickly a separation between the particulate materials in the body fluid and the remaining of the fluid. It is to be noted that the process equally may be used for body fluids from which the particulate materials have been removed already, f.i. plasma or serum, urine etc. because of the hydrophilic properties of the microporous membranes the liquid part of the body fluid penetrates easily and quickly into the pores, wherein such an amount of fluid is taken up that the pores are filled up. In the prior art generally use is made of hydrophobic membranes or films or paper discs which have been made hydrophobic. Because in the present invention one or more membranes are used, of which at least one (the collector membrane) has a defined pore volume, it is possible to take up into the membrane a properly determinable defined amount of fluid. According to the present process two objects are achieved simultaneously, to wit the separation of the particulate materials and the quantification of the absorbed fluid. This quantified amount of fluid may then be used according to two manners for the chemical analysis, to wit a dry and a wet analysis, which will be illustrated in more detail hereinafter. In particular the wet analysis is very advantageous, because said analysis enables to obtain quantitative results with the use of relatively cheap apparatus. Since the pore volume of the hydrophilic microporous membranes in the present process may be determined accurately, it is possible to obtain simply a quantitative result also in the dry analysis without special measures in order to obtain a defined amount of fluid sample. As mentioned before in the prior art generally use is made of swellable membranes or films, so that f.i. the body fluid should penetrate into the layer during a certain time in order to obtain an accurately determinable amount of absorbed fluid.

Since the total plasma/serum is recovered, it is according to the present process in principle possible to analyse each plasma/serum component. Some of the not limitative examples are : glucose, urea, cholesterol, proteins, lipides etc.

It should be understood that in the plasmapheresis, where the plasma is separated from whole blood, use is made of microporous membranes. However this process relates to the recovery of greater amounts, wherein the plasma is pressed through the membrane under pressure. The microporous membranes, which are used in the plasmapheresis, are in the device in general already humidified which means that said membranes do not take up the plasma in a quantitative amount without any pressure.

As it will be clear from the above, in principle any hydrophilic microporous membrane may be used, provided that it has a defined pore volume. A category of membranes which appeared to be suitable in the process according to the invention is disclosed in the older not-prepublished European patent application 87201789.2. The hydrophilic microporous membranes disclosed in the above European patent application have the considerable advantage that the pore structure may be adjusted accurately, which means that the membranes may be tailored.

According to a preferred embodiment of the process of the invention a hydrophilic microporous membrane is used, which essentially consists of a hydrophobic polymer and a hydrophilic polymer, which hydrophilic polymer is fixed in or on the polymer matrix. As described in the European patent application 87201789.2 this fixation may be achieved by cross-linking the hydrophilic polymer in an essentially not-swollen state. For further details relating to the preparation of suitable membranes, reference is made to this European patent application.

It is particularly preferred an embodiment of the present process wherein a membrane is used in which the hydrophobic polymer is represented by polysulphone, polyether sulphone or polyetherimide, whereas

the hydrophilic polymer is polyvinyl pyrrolidine.

A further preferred embodiment of the present process is characterized in that, after applying the slight amount of body fluid that part of the collector membrane to which/wherein the remained particulate materials are present and to which /wherein the non-absorbed fluid is present, is separated. By making use of a collector membrane, which may be separated in two zones, of which one zone contains the fluid residues and the other zone contains an accurately measurable of the absorbed fluid, the present process may be carried out by inserting the zone intended for the fluid residue in the body fluid to be analysed (f.i. blood or plasma or serum which is being received in a tube). The absorption zone and the fluid residue zone may be separated from each other. In order to accomplish this separation several methods are available which will be clear for the expert in the art.

A specially preferred embodiment of the process of the invention is characterized in that the slight amount of body fluid is applied onto a first surface of an additional hydrophilic microporous membrane (separator membrane), which is in contact with its opposite surface with the collector membrane in a separable way, wherein the body fluid penetrates in such a way that the collector membrane or an accurately determined part thereof is saturated, whereupon the separator membrane together with the not-absorbed fluid on it and eventually the particulate material remained on it and/or therein are removed. In such a process use is made therefore of two membranes which are in mutual contact with each other, of which one, to wit the separator membrane, retains the particulate materials and the not-absorbed fluid, whereas the other, to wit the collector membrane, is in contact with and absorbs the body fluid which is freed from the particulate materials by means of the separator membrane. However it is a condition for obtaining good results that the separator membrane and the collector membrane are in intimate contact with each other, f.i. by slightly pressing both membranes to each other. The time period which is required for the saturation of the collector membrane or an accurately determined part thereof varies between a few seconds to a few minutes dependent on the composition and structure of the membranes. The manner of how to contact the membranes with each other and the separation of these membranes from each other later is not critical and the various manners to achieve this will be clear to the expert.

Although symmetrical hydrophilic microporous membranes are suitable in the present process it is preferred to use at least one asymmetric hydrophilic microporous membrane. Under an asymmetric membrane should be understood a membrane wherein the average pore size at one side of said membrane is larger than the average pore size at the opposite side of said membrane. The techniques for the preparation of asymmetric membranes are known in itself and they also may be used for the preparation of asymmetric hydrophilic microporous membranes. Asymmetric membranes have been used up till now with their relatively smooth side, having the smaller pores, directed to the fluid to be separated or purified. However according to the invention the slight amount of body fluid is applied on that side of the asymmetric membrane which contains the largest pores. The advantage of this is that the body fluid together with the particulate materials if any penetrate immediately into the pores, wherein the particulate materials are retained in the pores which are becoming gradually smaller, whereas the clear part of the body fluid penetrates quickly further. As disclosed above the particulate materials which are retained in and/or on the membrane may not be washed off or wiped off, rather it is to be removed together with that part of the membrane which is contaminated with the particulate materials (in case that a wet analysis is to be carried out). Since the fluid is spread out to all directions because of the open pore structure (which does not occur with the particulate materials), it is possible to separate that part of the membrane wherein the particulate materials are retained (that is where the membrane came into contact with the body fluid). A very suitable alternative of this process according to the invention is characterized in that an asymmetric separator membrane is used, wherein the surface having the smaller pore openings is in contact with the collector membrane. Accordingly in this variant two membranes are used from which the separator membrane which is used for the application of the body fluid is asymmetric. In this way the separator membrane is to be separated from the collector membrane instead of splitting up a collector membrane into two parts. In this last mentioned preferred embodiment it is very favourable to make use of an asymmetric collector membrane too, from which the side with the smaller pore openings is in contact with the separator membrane. In this way the capillary suction of the clear body fluid into the collector membrane may occur optimally. It is self-explanatory that also several combinations of the here mentioned embodiments may be combined.

A further advantage of the present process lies in the possibility to provide the hydrophilic microporous membrane(s) with one or more reagents, which may be divided over one or more membranes. Accordingly one component which is to be analysed may be contacted with a reagent while in the membrane. It is preferred that the component(s) in the body fluid to be determined is/are reacted with the reagent(s) in or on the collector membrane. Said membrane in which the reagent containing body fluid is absorbed may be

used according to one variant directly for a dry analysis. The change in the physical characteristics which is a result of the reaction is measured according to a method which is known in itself. This may occur f.i. visually, leading to a semi-quantitative result, or by means of a suitable measuring device such as remission or reflection photometers, leading to a quantitative result. The second variant will be illustrated hereunder.

5 As discussed in the introductory part the present process has the special advantage that the same is suitable for a wet chemical analysis of components in a body fluid. In this connection the present process is characterized in that the collector membrane or an accurately determined separable part thereof, together with the body fluid absorbed therein is introduced into a suitable reagent solution, the fluid which is absorbed in the membrane(part) is washed out, and the changed physical characteristics caused by the  
10 reaction are measured in a way which is known in itself. In this case it is possible to incubate the absorbed fluid first with a certain reagent, which is f.i. collected in the collector membrane, as illustrated above. The absorbed fluid, eventually together with the reagent, is introduced into a suitable reagent solution after being washed out. In this way even a wet chemical analysis may be carried out, which consists of more than one separate steps. The reaction parameter, f.i. a colour change, may then be simply measured by means of a  
15 cheap apparatus. Because of the fact that hydrophilic microporous membranes according to the invention may be washed out in the reagent solution quantitatively, it is possible to obtain very accurate results. A wet analysis of body fluid which is carried out in this way has not been disclosed in the literature as yet. However such a method is of very high value, because it may be carried out quickly and simply. It will be clear that the fluid which is collected into the collector membrane (part) may be freed from its particulate  
20 materials by means of the above disclosed manners.

When the present process is applied for the chemical analysis of components of blood, it may be advantageous to use a hydrophilic microporous membrane, which contains coagulation- and/or haemolysis inhibiting substances. As examples of coagulation inhibiting substances may be mentioned heparine and ethylene diamine tetra acetic acid. Examples of haemolysis inhibiting substances are disclosed in U.S.  
25 patent specification 3,552,928. These coagulation-and haemolysis inhibiting substances may be introduced into the membrane by means of methods which are known in itself.

According to a second feature the present invention relates to a device for the separation of a body fluid from the particulate materials present therein, which device comprises an inert solid support to which a membrane or composite of membranes is attached, characterized in that said device is provided with one  
30 or more hydrophilic microporous membranes, at least one of which (collector membrane) has a defined pore volume. Such a device or strip is particularly suitable for carrying out the above disclosed process of the invention. The term inert solid support means a support which is manufactured from a material which is inert with respect to the or each membrane and which is furthermore inert with respect to the body fluid. Examples of such materials are the plastics and metals known for such applications. The shape of such a  
35 support is not critical, provided that during use the body fluid has to be able to come in contact with the or each of the membranes. Dependent on the situation whether a dry or wet chemical analysis is to be carried out, the membrane may be in contact with its one side with the support or the membrane should be free on both sides for absorption and washing out purposes. In case that the membrane which is attached to the support should be washed out, it is necessary that the support is inert too, with respect to the reagent  
40 solution.

As mentioned before the membrane may be a hydrophilic microporous membrane, provided that this has a defined pore volume. Suitable membranes consist essentially of a hydrophobic polymer and a hydrophilic polymer, which hydrophilic polymer is fixed in or onto the polymer matrix. In this connection the hydrophobic polymer is preferably polysulphone, polyether sulphone or polyetherimide, whereas the  
45 hydrophilic polymer is polyvinyl pyrrolidone.

According to a preferred embodiment of the device according to the invention such device comprises besides the collector membrane furthermore a hydrophilic microporous membrane (separator membrane), which is in a separable contact with the collector membrane. The advantage of such a device is that because of the simple removal of that part of the support to which the separator membrane is attached, the  
50 separation between the fluid residue and the accurately determined absorbed amount in the collector membrane may be accomplished.

A further advantageous embodiment of the present device is characterized in that said device comprises an asymmetric hydrophilic microporous membrane. The advantages of such a membrane are mentioned in detail hereinbefore. The pore size of the asymmetric membrane at the side with the largest  
55 pores is in the range of 10  $\mu\text{m}$  to 1 mm, whereas the pore size at the side with the smallest pores is in the range of 0.2 to 5  $\mu\text{m}$ . It is in particular advantageous that by introducing the body fluid at the side having the largest pores, said fluid penetrates very quickly into the membrane, whereas the particulate materials are partly carried with the fluid, which particulate materials finally are retained by the decreasing size of the

pores.

A device according to the invention which comprises a separator membrane and a collector membrane is preferably provided with an asymmetric separator membrane which is in contact with the collector membrane with its smallest pores. Such a device is furthermore preferably provided with an asymmetric collector membrane too, which is in contact with the separator membrane by its side having the smallest pores. The advantages of such embodiments of the present devices are illustrated with respect to the present process in more detail hereinbefore.

Another preferred embodiment of the device according to the invention is characterized in that the collector membrane is provided with an accurately determined separable part. This is advantageous too when the collector membrane is used together with a separator membrane. It should be understood that because of the removal an accurately determined separable part of the collector membrane, having a known pore volume, an accurate amount of absorbed fluid may be introduced into for instance a reagent solution, which enables the achievement of a quantitative measuring result. An additional advantage is that it is not required to fill up the entire membrane. The way of ascertaining of the separable part as well as the way of removal thereof from the rest of the collector membrane, may be carried out in different manners, so that a detailed explanation will not be given. One of these possibilities will be discussed later.

Another embodiment of the device according to the invention is characterized in that the collector membrane or the accurately determined separable part thereof is attached to an inert means which may be separated from the inert solid support. This attachment may be achieved by means of clamps, adhesives etc. Since the separable inert means is removable from the inert solid support, the collector membrane or a part thereof may be isolated in a simple way and used further. It is preferred that the separable means does clamp the collector membrane as well as optionally the separator membrane in or onto the inert solid support. Especially when both the collector membrane and the separator membrane are present, the contact between both membranes may be accomplished easily in this way.

As disclosed herein above with respect to the present process the hydrophilic microporous membrane contains a coagulation or haemolysis inhibiting substance in case that the device is to be used for chemical analysis of whole blood.

According to a final feature the present invention relates to a test kit for the separation and chemical analysis of the components of a body fluid, wherein the complete kit comprises one or more devices according to the present invention, one or more solid or liquid reagents in one or more suitable containers and optionally one or more solvents in separate containers. When a fluid residue of a membrane has to be removed it is preferred that said test kit also comprises a suitably packed isotonic solution which is present in a tissue in a saturated or not saturated state.

The terms collector membrane and separator membrane as used in the specification and claims refer firstly to the functions of the used membranes. The term collector membrane refers to the membrane having a defined pore volume and which contains the amount of fluid to be analysed. The term separator membrane is only used in order to indicate the use of an additional membrane, applied to a collector membrane, wherein the main function of the separator membrane is to retain the fluid residue which is not intended for chemical analysis. In case that only one membrane is used, this is indicated as collector membrane, although this membrane also has the function to separate the particulate materials from the rest of the fluid.

The invention will be now further illustrated by means of the embodiments as illustrated in the enclosed drawings of two testing devices according to the invention, to which the invention is not limited. In the drawings:

Fig. 1 shows a perspective view of a first embodiment of the present device;

Fig. 2 is a partial sectional view according to the plane II-II of the embodiment of the device according to fig. 1 in exploded view;

Fig. 3 shows a perspective view of a second embodiment of the present device;

Fig. 4 is a partial sectional view according to the plane IV-IV of the embodiment of fig. 3 in a fold-out position.

In fig. 1 an embodiment of the device for the separation of a body fluid from particulate materials in said fluid, in particular plasma from whole blood, is indicated by the reference figure 1. This embodiment is provided with a handle 2 and means 3 for receiving the separator-and collector membrane. The separator membrane in fig. 1 is indicated by the reference figure 5, whereas the collector membrane 7 in this figure is not shown. Finally figure 6 indicates the handle of an inert removable means.

By referring to figures 1 and 2, which show a partial sectional view according to the plane II-II of fig. 1 in exploded view, as mentioned before, the separator membrane 5 and the collector membrane 7 may be



inserted into the space 14 of the means 3 of the solid support 1. The figure 16 indicates the other end of the removable means, which is connected with the handle 6. This means 16 is provided with a grate 10, which is breakably connected with the remaining part 11. Said means 16 may be inserted together with the membranes 5 and 7 into the space 15 of the support 1, f.i. by means of a snap-closing, which enables that both membranes are pressed against the grate 4, so that between said membranes a suitable contact occurs. The reference figure 13 indicates a weak spot which enables to remove the grate 10 of the remaining part 11; f.e. when the handle 6 of the removable means in fig. 1 is moved in an upward direction. The collector membrane 7 comprises in this embodiment a zone 12, the size of which has been determined accurately and a surrounding zone 9, which both zones are connected to each other by means of a weak binding 8. By attaching zone 12 to the grate 10 and by moving the handle 6 in upward direction the zone 12 will be separated from the zone 9 by breaking along the weak spots. The handle 6 with the grate 10 and the zone 12 of the collector membrane attached thereto may be removed and used for chemical analysis. The collector membrane zone 9 as well as part 11 remain in part 3 of the support 1. The shape of the membranes 7 and 5 may be circular, square etcetera, whereas the weak spots 8 and 13 may be circular or right-angular. The weakening parts 8 and 13 may be obtained in several manners, a discussion of which may be avoided.

Fig. 3 shows a second embodiment of the device according to the present invention, which is indicated by the reference figure 21, wherein 22 and 23 are handles and 24 and 25 represent rooms for receiving the membranes. Figure 28 represents a hinge, which enables to fold out this device by moving the handles 22 and 23 away from each other (see fig. 4). The separator membrane is indicated by 27 and is kept in position by means of grate 26.

Fig. 4 is a partial sectional view according to the plane IV-IV of the device of fig. 3, wherein said device is folded out. In order to ensure a proper contact between the separator membrane 27 and the collector membrane 29 snap-closings 31, 34 respectively 32, 33 are provided. The separator membrane 29 is similarly supported by a grate 30. The left part of this embodiment in fig. 4 may be separated from the right part by breaking or loosening of the hinge 28. The left part may then be further used for chemical analysis.

It should be understood that the above disclosed embodiments of the device according to the present invention only serve as example and that from that embodiment several alternatives may be developed which are also covered by the present invention.

The invention will be further illustrated by means of the following examples, to which the invention is not limited.

#### Example I

In order to determine the contents and reproducibility of the hydrophilic microporous membranes of the invention a known amount of cholesterol calibration serum type A of the R.I.V.M. of Bilthoven having a cholesterol concentration of 3.68 mmol/l is applied on pieces of membranes, followed by waiting a few seconds until the liquid was penetrated into the membrane. Then the humidified surface, which appeared to be the same on both sides of the membrane, was measured. In order to determine the influence of the type of liquid, in particular the influence of the viscosity, the same test has been carried out with water.

The experiments were carried out with membranes of the type PS-11, obtainable from the firm X-FLOW B.V. at Enschede, which membranes were composed of polyether sulphone and polyvinyl pyrrolidone having a pore size of about 0.5 micrometer. The thickness of the membrane was 0.2 mm. The results are listed in table A hereunder.



TABLE A

Test liquid	amount ( $\mu$ l)	surface (mm <sup>2</sup> )		
		measurement 1	measurement 2	measurement 3
a. Testserum	5	49	50	49
	10	101	101	99
	15	150	148	151
B. Water	5	50	51	49
	10	100	99	99
	15	152	148	150

From this example it appears that the porosity of the membranes, and therefore the amount of plasma which may be absorbed, are sufficiently reproducible for carrying out quantitative measurements.

#### Example II

A. Onto a hydrophilic microporous membrane of the type PS-11, see example I, having dimensions of 7 mm x 25 mm on one side thereof there is applied a drop of blood obtained from a 45 year old man by finger puncture. After 1 minute a piece of membrane having dimensions of 7 mm x 7 mm was removed therefrom by cutting, which piece was saturated with plasma. The membrane part to which the drop of blood was applied, was discarded.

The piece of membrane of 7 mm x 7 mm was transferred into a cuvet which contained 1 ml of cholesterol reagent of the type Medela E 550-R, wherein the contents of said membrane has been washed out.

The test was carried out according to the known enzymatic coloric method at a wave length of 540 nm in a Vital Scientific Vitalab 10 colorimeter, which was calibrated by means of a standard cholesterol-calibration serum of 3.68 mmol/l and 8.01 mmol/l, derived from R.I.V.M. under type A.

The cholesterol content of said blood was determined on a value of 5.2 mmol/l at a contents of the membrane of 5 microlitres.

B. As a control of the test under A hereinabove blood was taken from the same person at the same moment by means of a vena puncture, followed by centrifugation of said blood and from the obtained serum there was 5  $\mu$ l added to 1 ml of the same cholesterol reagent. Then the cholesterol contents of the serum has been determined similarly as disclosed in A above. The colorimetric measurement showed a cholesterol contents of 5.2 mmol/l.

C. From the serum obtained under B a drop has been applied to a membrane PS-11 having a dimension of 7 x 25 mm. This has been treated similarly as under A above. The measurement showed a cholesterol contents of 5.3 mmol/l.

#### Example III

A. An asymmetric hydrophilic microfiltration membrane of the type PS-21, available at the firm X-FLOW B.V. at Enschede, which membrane essentially consisted of polyether sulphone and polyvinyl pyrrolidone and on one side it had a pore size of 0.5  $\mu$ m and at the opposite side a pore size varying between 50 and 100  $\mu$ m, and having a thickness of 0.5 mm, with its side having the smallest pores has been applied to a membrane of the type PS-11, which had a dimension of 11 mm x 11 mm.

Onto the membrane PS-21 about 50  $\mu$ l whole blood, obtained by finger puncture from a 38 year old man was applied. The particulate materials remained in the membrane PS-21, whereas the plasma penetrated very quickly (within a few seconds) in the membrane PS-11 thereunder. Then the membrane PS-21 has been removed and the membrane PS-11 was transferred into a cuvet containing 1 mm cholesterol reagent of the type Medela E 550-R, wherein the membrane was washed out.

At a membrane contents of 12  $\mu$ l, the cholesterol content of the blood was determined on a value of 5.8 mmol/l.

B. As control 12  $\mu$ l serum, derived from centrifugated blood, which was obtained by means of a vena

puncture with the same person at the same moment, was directly added to a cuvet which contained 1 ml of the cholesterol reagent. This control test resulted similarly in a cholesterol content of 5.8 mmol/l.

## 5 Claims

1. A process for the separation of a body fluid from particulate materials in said fluid, in particular plasma from whole blood, by contacting a small amount of said body fluid respectively whole blood with a membrane or composite of membranes, wherein the body fluid respectively the whole blood penetrates into the membrane or composite of membranes, wherein the particulate materials and non-penetrated body fluid respectively whole blood penetrates, whereas the penetrated fluid respectively plasma is used for analysis purposes, characterized in that one or more hydrophilic microporous membranes, at least one of which (the collector membrane) has a defined pore volume, is or are used, wherein the absorbed body fluid, freed from the particulate materials if any and having a defined amount is used for a semi-quantitative or quantitative, dry or wet chemical analysis for the determination of one or more components thereof.

2. The process according to claim 1, characterized in that a hydrophilic microporous membrane is used, which essentially consists of a hydrophobic polymer and a hydrophilic polymer, which hydrophilic polymer is fixed in or on the polymer matrix.

3. The process according to claim 2, characterized in that a membrane is used, wherein the hydrophobic polymer is polysulphone, polyether sulphone or polyetherimide, whereas the hydrophilic polymer is polyvinyl pyrrolidone.

4. The process according to claims 1-3, characterized in that after applying said small amount of body fluid onto the membrane or composite of membranes that part of the collector membrane whereon/wherein the remaining particulate materials are present and whereon/wherein the non-absorbed fluid is present, is separated.

5. The process according to claims 1-3, characterized in that said small amount of body fluid respectively whole blood is applied to a first surface of another hydrophilic, microporous membrane (separator membrane), which with its opposite surface is in breakable contact with the collector membrane, one allows the body fluid respectively whole blood to penetrate into the collector membrane until an accurately determined part thereof is saturated, whereupon the separator membrane together with the non-absorbed fluid respectively whole blood and the remained particulate materials present thereon is removed.

6. The process according to claims 1-5, characterized in that at least asymmetric hydrophilic microporous membrane is used.

7. The process according to claim 6, characterized in that said small amount of body fluid respectively whole blood is applied to that side of the asymmetric membrane having the largest pores.

8. The process according to claim 5-7, characterized in that an asymmetric separator membrane is used, which is in contact with the collector membrane with its surface having the smallest pore sizes.

9. The process according to claim 8, characterized in that an asymmetric collector membrane is used which is in contact with the separator membrane with its surface having the smallest pore size.

10. The process according to claims 1-9, characterized in that the component(s) in the absorbed body fluid respectively plasma to be determined is (are) reacted with the reagent(s) applied in or on said collector membrane.

11. The process according to claim 10, characterized in that the physical characteristics which have been changed due to the reaction are determined in a way which is known in itself.

12. The process according to claims 1-10, characterized in that said collector membrane or an accurately determined separatable part thereof, together with the therein absorbed body fluid respectively plasma is transferred into a suitable reagent solution, the fluid respectively plasma absorbed into the membrane or part of the membrane is washed out, whereas the physical characteristics changed due to the reaction are determined in a way which is known in itself.

13. The process for the chemical analysis of components of whole blood according to claims 1-12, characterized in that a hydrophilic microporous membrane is used containing coagulation and/or haemolysis inhibiting substances.

14. A device for the separation of a body fluid from particulate materials in said fluid, in particular plasma from whole blood, to be used for the chemical analysis comprising an inert solid support to which a membrane or composite of membranes is attached, characterized in that said device is provided with one or more hydrophilic microporous membranes from which at least one (collector membrane) has a defined pore volume.

15. Device according to claim 14, characterized in that each of the hydrophilic microporous membranes essentially consists of a hydrophobic polymer and a hydrophilic polymer, which hydrophilic polymer is fixed in or onto the polymer matrix.

16. Device according to claim 15, characterized in that said hydrophobic polymer is polysulphone, polyether sulphone or polyetherimide, whereas the hydrophilic polymer is polyvinyl pyrrolidone.

17. Device according to claims 14-16, characterized in that said device besides said collector membrane also comprises another hydrophilic microporous membrane (separator membrane), which is in breakable contact with said collector membrane.

18. Device according to claims 14-17, characterized in that said device is provided with an asymmetric hydrophilic microporous membrane.

19. Device according to claim 18, characterized in that the pore size of the asymmetric membrane on its surface having the largest pores is in the range of 10  $\mu\text{m}$  to 1 mm, whereas the pore size at the surface having the smallest pores is in the range of 0.2 to 5  $\mu\text{m}$ .

20. Device according to claims 17-19, characterized in that the separator membrane is assymmetric and is in contact with the collector membrane by means of its surface having the smallest pores.

21. Device according to claim 20, characterized in that said collector membrane is also asymmetric and is in contact with said separator membrane by its surface having the smallest pores.

22. Device according to claims 14-21, characterized in that said collector membrane has an accurately determined removable part.

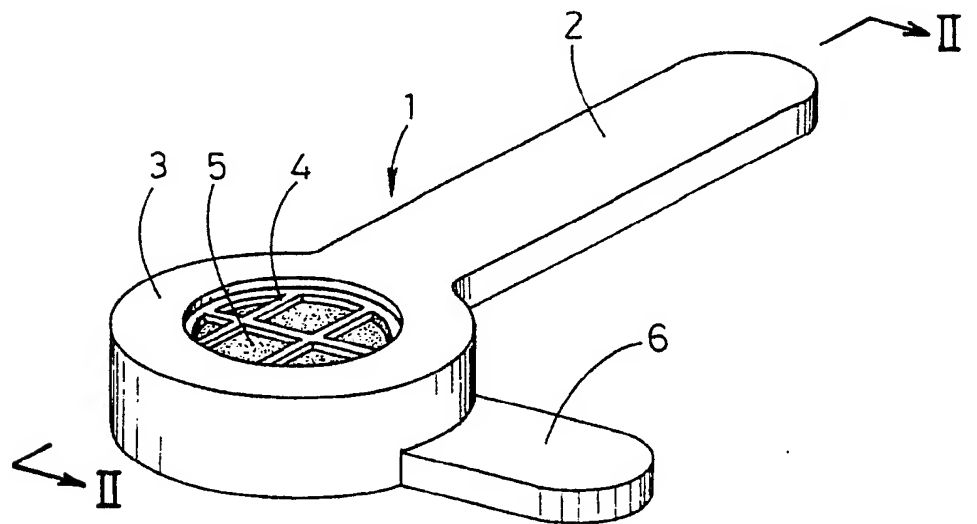
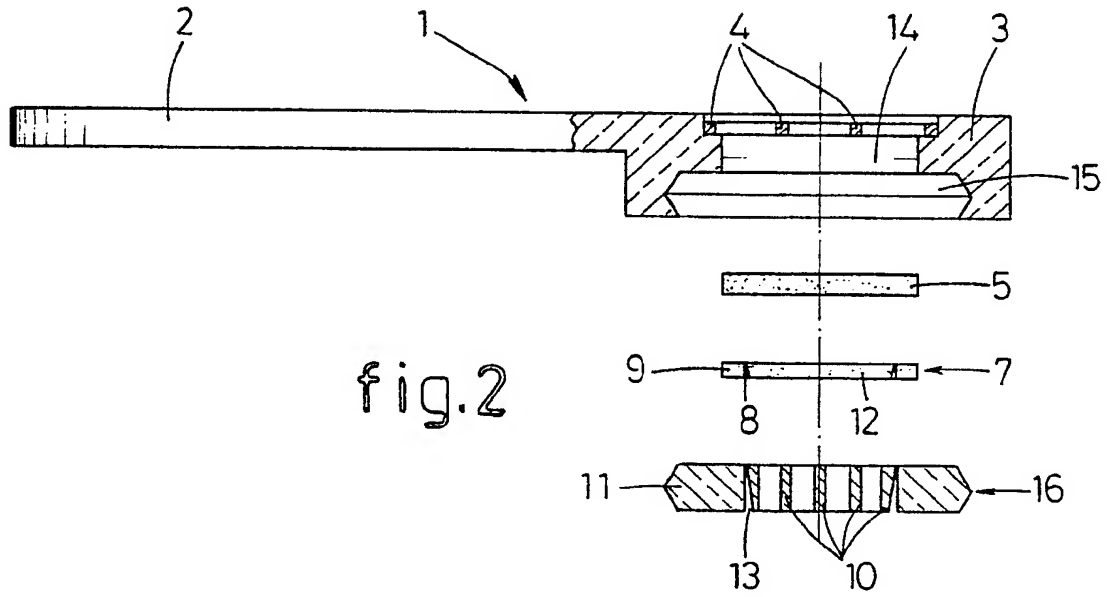
23. Device according to claims 14-22, characterized in that said collector membrane or the accurately determined removable part thereof is attached onto an inert means which is removable from the inert solid support.

24. Device according to claims 23, characterized in that said removable means, said collector membrane and said separator membrane if any clamps onto the inert solid support.

25. Device for the separation respectively the chemical analysis of whole blood according to claims 14-25, characterized in that said hydrophilic microporous membrane contains a coagulation or haemolysis inhibiting substance.

26. Testing kit for the separation of a body fluid from particulate materials in said fluid, in particular plasma from whole blood and for the chemical analysis for the components therein, characterized in that said testing kit is provided with one or more devices according to claims 14-25, one or more solid or liquid reagents contained in one or more suitable containers and optionally one or more solvents contained in one or more separate containers.

27. Testing kit according to claim 26, further characterized in that it is provided with a suitably packed isotonic solution, which is present in a saturated or unsaturated state in a tissue.



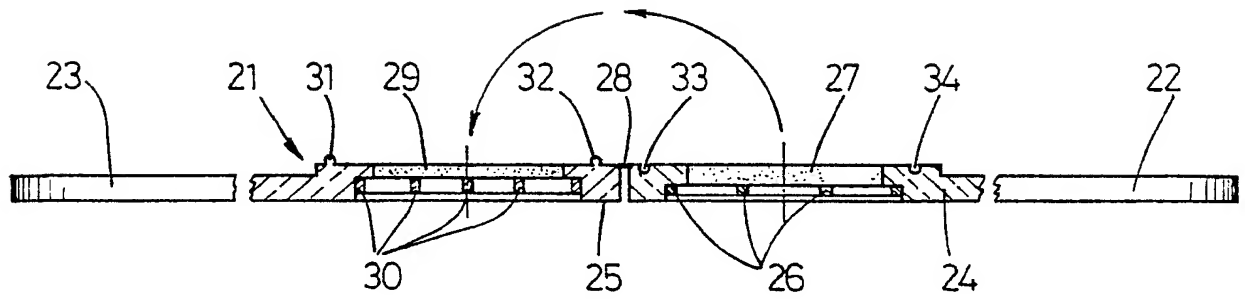


fig.4

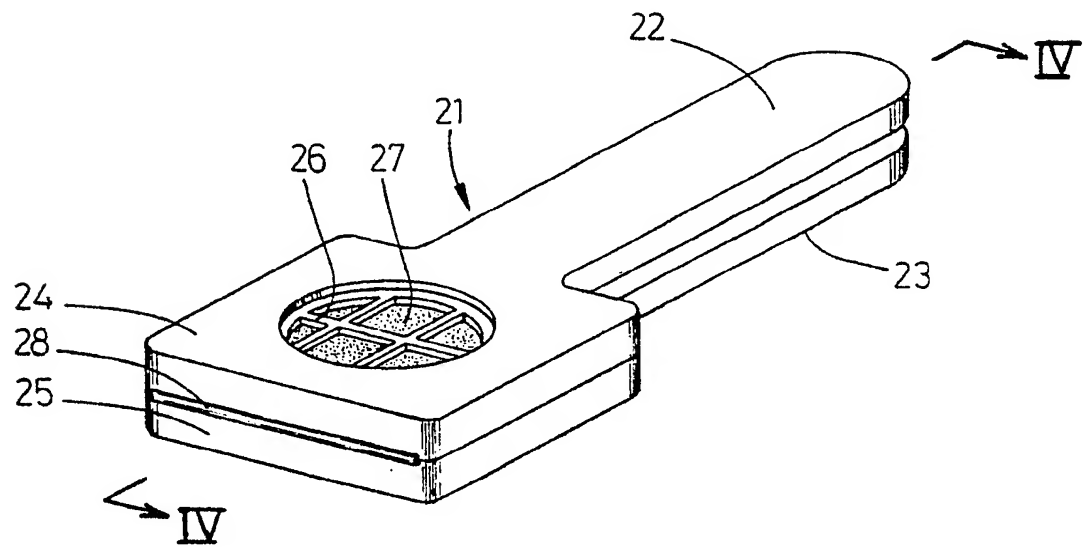


fig.3



EP 89 20 0767

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.4)
X	PATENT ABSTRACTS OF JAPAN, vol. 11, no. 233 (P-600)[2680], 30th July 1987; & JP-A-62 46 260 (KONISHIROKU PHOTO IND. CO., LTD) 28-02-1987 * Abstract * ---	1,2,15,16	G 01 N 33/52 G 01 N 33/48
X	EP-A-0 154 839 (BAYER AG) * Abstract; claims 1,4,8,9; page 32, example 5; page 20, lines 18-25; page 21, lines 1-28; page 17; page 6, lines 1-26 * ---	1-3,6,7,15-17,19	
X	EP-A-0 082 433 (HOECHST AG) * Abstract; claims * ---		
X	EP-A-0 175 990 (BAYER AG) * Abstract; claims 1-8; page 26, lines 13-28; page 27, lines 1-10 * ---	1-5,15-19	
D,E	EP-A-0 261 734 (X-FLOW B.V.) * Pages 2,3; columns 1-3; page 4 * ---	2,3,16,17	TECHNICAL FIELDS SEARCHED (Int. Cl.4)
A	EP-A-0 217 403 (ABBOTT) * Figures 1,2,5; page 7, column 12, lines 5-53 * ---	15-25	G 01 N B 01 D C 12 Q
A	PATENT ABSTRACTS OF JAPAN, vol. 12, no. 185 (P-710)[3032], 31st May 1988, page 34 P 710; & JP-A-62 293 161 (KONISHIROKU PHOTO IND. CO., LTD) 19-12-1987 * Abstract * ---	15-25	
A	US-A-3 791 933 (R.H. MOYER et al.) -----		
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 05-07-1989	Examiner MEYLAERTS H.
<b>CATEGORY OF CITED DOCUMENTS</b>			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons ----- & : member of the same patent family, corresponding document	

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